Principle 2

Principle 2: The ocean and life in the ocean shape the features of Earth.

All matter on Earth cycles through various reservoirs: the atmosphere, hydrosphere, biosphere, and lithosphere.

Rock Cycle and Plate Tectonics — A					Biogeochemical Cycles — B			
All the rocks on land will end up in the ocean due to weathering and erosion. The continual formation and breakdown of rocks constitutes the rock cycle.				rock cycle.	The ocean plays a major role in the biogeochemical cycles that are fundamental to life on Earth.			
A1 A7					B1			
Rocks are constantly being broken down and recycled through weathering, erosion, and processes associated with plate tectonics, such as subduction and uplift.Rocks are constantly being formed through accretion, sedimentation, volcanism, and igneous processes.					All elements are present in ocean water at various concentrations. Many elements in the ocean are needed by all living organisms. These include C, P, N, S, O and many metals such as Fe, Zn, Ca, Na, K. Other elements (Si, Sr) are needed by some select organisms.			
A2	A3	A8	A9	A10	Carbon Cycle — B2	Phosphorus Cycle — B12	Nitrogen Cycle — B19	Silica Cycle — B26
Many products of weathering and erosion enter the ocean via rivers and atmospheric depo- sition. All matter remains in the ocean for different lengths of time (residence times).	subducted beneath continental plates when the two collide, causing the	Accretion is the process by which material is added to a tectonic plate through subduction and uplift, sea level change, and wave action.	Sedimentation in the ocean can occur by the process of material settling out of the water by gravity so it accumulates on the sea floor, or by the process of currents or waves moving material along the sea floor.	sedimentary rocks, creates	The ocean is the largest reservoir of rapidly cycling organic and inorganic carbon on Earth.	All life on Earth depends on phosphorus for important compounds, e.g., ATP, DNA, and phospholipids.	All life on Earth depends on nitrogen for amino acids and proteins. Most of the nitrogen (N) on Earth is in the atmosphere as N ₂ , which cannot be used directly by most organisms.	Some oceanic organisms (e.g., diatoms, radiolarian, sponges) use silica to construct the hard parts of their body, such as tests, frustules, spines, and spicules.
	A4			·	B3	B13	B20	B27
	Subduction can result in the addition of oceanic rocks and sediments to the upper mantle or to the edge of the continent.				Carbon in the form of carbon dioxide in the atmosphere enters the ocean by diffusion, convective mixing, and bubble entrainment. Carbon is present in the ocean in dissolved inorganic (CO ₂ , H ₂ CO ₃ , HCO ₃ , CO ₃ , CH ₄), organic (carbohydrates, lipids, amino acids), and in particulate forms (carbonate minerals and living and dead organisms	Terrestrial weathering of rocks is the primary source of phosphorus (P) in the ocean.	Some bacteria (including some cyanobacteria) in the ocean can fix N_2 in the atmosphere and convert it to organic forms at the cellular level, thus making nitrogen available to other marine organisms.	Terrestrial weathering of rocks and volcani eruptions are the primary sources of silica (Si) in the ocean. Rivers transport the silica weathered from rocks on land to the ocean.
	A5 A6				B4 B7	B14	B21	B28
	Ocean trenches, island arcs, stratovolcanoes, and some mountainSome parts of the ocean (e.g. the Pacific Rim) are dominated by subducted plate boundaries.				Inorganic carbon is converted by photo- and chemoautotrophs to organ- ic matter in the process of carbon fixation (e.g., photo- and chemosynthesis.Inorganic carbon dissolved i seawater is used by some or- ganisms, such as corals, prot zoa, and mollusks, to form calcium carbonate (CaCO3) shells and other skeletal part	dissolved inorganic (HPO ₄ , PO ₄) and organic o- forms (phospholipids, sugars) as well as in particulate forms (e.g., minerals, such as apatite), and living and dead organisms.	Nitrogen (N) is continuously transformed between various oxidation states (NO ₃ , NO ₂ , NH ₄) and compounds (amino acids, urea) in the ocean through biologically- mediated transformation processes, such as nitrogen fixation, nitrification, denitrification, assimilation, ammonification, and anaerobic ammonium oxidation.	Silica (Si) is present in the ocean in dissolved inorganic (SiO ₂) and particulate forms (clays and other materials from land and biogenic skeletal materials).
					B5 B8	B15	B22	B29
					Most of the organic carbon is converted back to inorganic carbon through the process of respiration. Much of the calcium carbonate from shells dissolves in the deep ocean.	 Phytoplankton and other primary producers take up phosphorus dissolved in seawater and convert it to biomass, which is consumed by heterotrophic organisms higher in the food chain. In the process of respiration and regeneration, dissolved and particulate organic phosphorus is converted back to dissolved inorganic forms. Upwelling brings inorganic phosphorus back to the surface. 	Phytoplankton and other primary producers take up ni- trogen dissolved in seawater and convert it to biomass, which is consumed by heterotrophic organisms higher in the food chain. In the process of respiration and re- generation, dissolved and particulate organic nitrogen is converted back to dissolved inorganic forms. Up- welling brings inorganic nitrogen back to the surface.	Most of the biogenic silica skeletal material is dissolved in the ocean after the organisms die, and is converted back to dissolved inorganic silica. Upwelling brings inorganic silica back to the surface.
					B6 B9	B16	B23	B30
					Some organic carbon sinks to the sea floorSome calcium carbonate from shells and corals accu- mulates to form sedimentar rocks, i.e., limestone, chalk, and carbonate banks.	y ocean sediments, where it undergoes transformations	Some organic forms of nitrogen (e.g., lipids, sugars) accumulate in ocean sediments, where they become part of the organic matter in sedimentary rocks.	Some relatively small fractions of the biogeni silica accumulates in ocean sediments, where it undergoes transformations and, over time, becomes part of sedimentary rocks, such as chert, diatomite, and clay materials.
					B10	B17	B24	B31
					Uplift and accretion processes, as well as sea level changes, may relocate sedimentary rocks containing both organic and inorganic carbon onto land, where rocks can undergo weathering and erosion before eventually returning to the ocean.	Uplift and accretion processes, as well as sea level changes may relocate oceanic sedimentary rocks containing both organic and inorganic phosphorus onto land, where the rocks can undergo weathering and erosion before eventually returning to the ocean.	Uplift and accretion processes, as well as sea level changes may relocate oceanic sedimentary rocks containing nitrogen onto land, where the rocks can undergo weathering and erosion before eventually returning to the ocean.	Uplift and accretion processes, as well as sea level changes, may relocate oceanic sedimen tary rocks containing silica onto land, where the rocks can undergo weathering and erosic before eventually returning to the ocean.
					B11	B18	B25	B32
Indbook for Increasing Ocear	Literacy				Anthropogenic burning of fossil fuels converts organic (petroleum) to inorganic carbon (CO ₂) and releases carbon back into the atmosphere, which affects the climate and pH balance of ocean water.	Anthropogenic loading of phosphorus to waterways, (e.g., rivers, runoff, lakes, and groundwater) from fertilizers and waste water reaches the ocean, where it becomes available to primary producers. Phosphorus may cause eutrophication (i.e., harmful algal blooms) and alter food webs.	Anthropogenic loading of nitrogen to waterways (e.g., rivers, runoff, lakes, and groundwater) and the atmosphere from fertilizers, waste water, and fossil fuel combustion, may reach the ocean, where it becomes available to primary producers. Nitrogen may cause eutrophication (i.e., harmful algal blooms) and alter food webs.	Anthropogenic activities that alter continental weathering rates (e.g., acid rain) and affect the global water cycle may change the flux of silica into the ocean.